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Stereoselective reduction of N-hydroxy-\alpha-iminocarbonyl-oligopeptide methyl esters with Zn-MsOH

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Abstract—The reduction of *N*-hydroxy- α -imino esters with Zn–MsOH in THF afforded α -amino esters in high yields. The reduction of *N*-hydroxy- α -iminocarbonyl-oligopeptide methyl esters prepared from L-phenylalanine and L-leucine di-, tri-, tetrapeptides gave the corresponding *S*-formed oligopeptide methyl esters in moderate diastereoselectivities. © 2002 Elsevier Science Ltd. All rights reserved.

Reduction of oximes is a useful method for the synthesis of amines from carbonyl compounds.¹ Recently, we have reported the reductive coupling of aromatic aldoximes and azines to N,N'-unsubstituted 1,2diamines with Zn-MsOH or Zn-TiCl₄.^{2,3} We found that the reduction with Zn-MsOH was also effective for the conversion of N-hydroxy- α -imino esters to α amino esters. We further investigated the reduction of N-hydroxy-a-imino carbonyl groups bonded to the Nterminus of L-phenylalanine or L-leucine oligopeptides with Zn-MsOH (Eq. (1)). Herein we report that the diastereoselectivity in the reduction of N-hydroxy-aiminocarbonyl-oligopeptide methyl esters was strongly affected by the number of amino acid residues of the oligopeptide moieties. The S-selectivity increased with increase in the number of amino acid residues (n) in the substrates, N-hydroxy- α -iminocarbonyl-(L-Phe)_n-OMe and N-hydroxy- α -iminocarbonyl-(L-Leu), -OMe (n=1-4).

HO, N
R¹ (CONH) CO₂Me
$$\xrightarrow{1) \text{Zn-MsOH}}$$
 R¹ (CONH) CO₂Me $\xrightarrow{1) \text{Zn-MsOH}}$ R¹ (CONH) CO₂Me $\xrightarrow{R^2}$ (1)
R¹ = Me, Bn, *i*-Bu; R² = Bn, *i*-Bu; n = 1-4 (1)

First, the reduction of N-hydroxy- α -imino esters was carried out with Zn–MsOH in THF at 25°C. After

N-Boc-protection of the resulting α -amino esters, *N*-Boc- α -amino esters were obtained in excellent yields (Eq. (2)). In order to complete the reduction, the addition of MsOH in amounts equimolar with Zn was required. These results show that Zn–MsOH is an effective reducing agent for the reduction of *N*-hydroxy- α -imino esters to α -amino esters.

HO, N

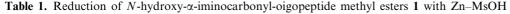
$$R^{1}$$
 CO₂Me $\xrightarrow{1) \text{Zn-MsOH}}$ $\xrightarrow{\text{NHBoc}}$
 R^{1} CO₂Me $\xrightarrow{R^{1}}$ CO₂Me $\xrightarrow{R^{1}}$ R¹ = Me, 87% $\xrightarrow{R^{1}}$ = Bn, 88% $\xrightarrow{R^{1}}$ = *i*-Bu, 90% (2)

Second, a series of N-hydroxy- α -iminocarbonyloligopeptide methyl esters 1a-d (R¹=Me) and 1e-h $(\mathbf{R}^1 = \mathbf{Bn})$ prepared from L-phenylalanine methyl ester (n=1) and L-phenylalanine oligopeptide methyl esters (n=2-4) were reduced with Zn-MsOH in THF or DMF at 25°C. The results are summarized in Table 1. In all cases, N-Boc-oligopeptide methyl esters 2 were obtained as mixtures of two diastereomers ((S)-2) and (R)-2) in good to excellent yields. The S:R ratios at the N-terminal residues in 2 were determined by comparison of their ¹H NMR spectra with those of authentic samples.⁴ Fig. 1 illustrates the correlation between the percentage of the S-form in 2 and the number of amino acid residues (n). In THF, the S-selectivity rose steadily with an increase of the number of residues. Whereas slight *R*-selectivities were found in the reduction of the substrates 1a and 1e derived from L-phenylalanine

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				Ph 1) Zn-Me / THF CO ₂ Me 2) (Boc) ₂ n / NaH	or DMF		L)	+		Boc CONH (<i>R</i>)-2	
In THF						In DMF					
1	\mathbb{R}^1	п		Yield ^a (%) of 2	S:R ^b	1	\mathbb{R}^1	n		Yield ^a (%) of 2	$S:R^{b}$
1a	Me	1	2a	91	45:55	1a	Me	1	2a	89	41:59
1b	Me	2	2b	85	68:32	1b	Me	2	2b	87	55:45
1c	Me	3	2c	77	75:25	1c	Me	3	2c	78	60:40
1d	Me	4	2d	71	85:15	1d	Me	4	2d	75	60:40
1e	Bn	1	2e	92	48:52	1e	Bn	1	2e	87	42:58
1f	Bn	2	2f	89	63:37	1f	Bn	2	2f	88	44:56
1g	Bn	3	2g	83	70:30	1g	Bn	3	2g	85	46:54
1h	Bn	4	2h	77	82:18	1ĥ	Bn	4	2h	66	42:58

^a Isolated yields.

^b Determined by ¹H NMR spectra.

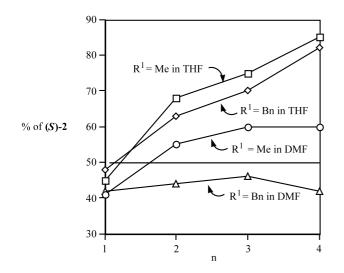


Figure 1. Correlation between percentage of (S)-2 and number of residues (n).

methyl ester (n=1), S-formed diastereomers of 2 were obtained preferentially from the substrates 1b-d and 1f-h made of L-phenylalanine di-, tri-, and tetrapeptides methyl esters (n=2-4). Unfortunately, the substrate prepared from a pentapeptide methyl ester (n=5) was sparingly soluble in THF. On the other hand, when the reduction of 1 was carried out in DMF, the stereoselectivities were relatively low (40-60% of S-2). Especially, in the cases of $\mathbf{R}^1 = \mathbf{Bn}$ (1f-h), the selectivity remained stable at 40-45% of the S-form. Next, TFA and TiCl₄ were used as an additive in place of MsOH. As depicted in Table 2, the yields of 2 were slightly less than those obtained with Zn-MsOH. As to the diastereoselectivity in 2, Zn-TFA gave similar results to those with Zn-MsOH, while $Zn-TiCl_4$ afforded considerably low Sstereoselectivities.

Another series of substrates 3a-d (R¹=Me) and 3e-h $(\mathbf{R}^1 = i - \mathbf{B}\mathbf{u})$ derived from L-leucine methyl ester (n = 1)and oligopeptide methyl esters (n=2-4) were also

_Ph

	HC Me	N L		Zn-additive / THF (Boc) ₂ O / NaHCO ₃ aq	NHВос 	n n	. J	HBoc CONH (<i>R</i>)-2		
		W	Vith Zn–TFA		With Zn–TiCl ₄					
1	п		Yield ^a (%) of 2	2 S:R ^b	1	п		Yield ^a (%) of 2	S:R ^b	
1a	1	2a	66	40:60	1a	1	2a	71	44:56	
1b	2	2b	63	67:33	1b	2	2b	69	40:60	
1c	3	2c	61	70:30	1c	3	2c	65	49:51	
1d	4	2d	55	76:24	1d	4	2d	48	46:54	

_Ph

Table 2. Reduction of 1a-d with Zn-TFA or Zn-TiCl₄ in THF _Ph

^a Isolated yields

^b Determined by ¹H NMR spectra.

HO

reduced with Zn–MsOH in THF at 25°C as exhibited in Table 3. Fig. 2 represents the correlation between the % of the S-form in 4 and n. Similarly to the reduction of 1, the S-forms of 4 were obtained selectively from the substrates **3b–d** and **3f–h** (n=2-4). However, the S-selectivity in 4 rose rapidly from n=1 to n=2 and leveled off practically at n=2.

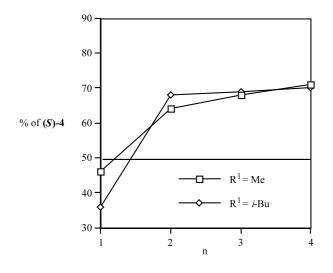
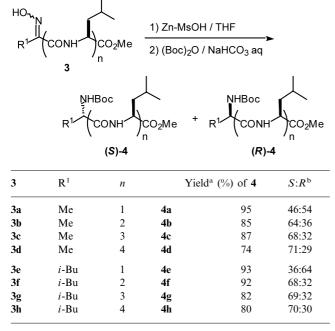


Figure 2. Correlation between percentage of (S)-4 and number of residues (n).

As another reducing agent, NaBH₄–CoCl₂· $6H_2O^5$ was employed for the reduction of 1 (Table 4). In contrast to the results obtained with Zn–MsOH, the diastereoselectivities in 2 were very low irrespective of the number of residues.

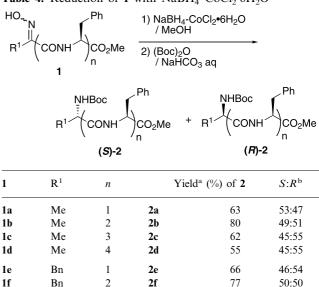
Table 3. Reduction of *N*-hydroxy- α -iminocarbonyloligopeptide methyl esters 3 with Zn–MsOH in THF



^a Isolated yields.

^b Determined by ¹H NMR spectra.

Table 4. Reduction of 1 with NaBH₄-CoCl₂·6H₂O



2g

2h

82

59

51:49

46:54

^a Isolated yields.

Bn

Bn

1g

1h

^b Determined by ¹H NMR spectra.

3

4

In conclusion, the reduction of *N*-hydroxy- α -imino esters and *N*-hydroxy- α -iminocarbonyl-oligopeptide methyl esters with Zn–MsOH in THF gave effectively the corresponding α -amino esters and oligopeptide methyl esters, respectively. The diastereoselectivity in the reduction of *N*-hydroxy- α -iminocarbonyl-oligopeptide methyl esters was not influenced largely by the substituents in *N*-hydroxy- α -iminocarbonyl groups (R¹) and oligopeptides (R²), but much affected by the number of residues in oligopeptides (*n*). Moderate *S*-selectivities were observed in the reduction of the substrates derived from L-phenylalanine and L-leucine di-, tri-, tetrapeptides (*n*=2–4).

General procedure is as follows: zinc powder (Wako Pure Chemical Industries, Ltd.) was washed with 1 M HCl, water, and ether successively and dried in vacuo. To a solution of N-hydroxy- α -iminocarbonyl-oligopeptide methyl ester 1 or 3 (1 mmol) in dry THF (5 mL for n=1and 2, 10 mL for n = 3, 20 mL for n = 4) was added freshly distilled MsOH (0.65 mL, 10 mmol) and zinc powder (0.65 g, 10 mmol) at 0°C under an atmosphere of nitrogen. The mixture was allowed to warm to 25°C and stirred for 8 h. To the mixture was added satd NaHCO₃ in water (20 mL) and (Boc)₂O (0.44 g, 2 mmol). The mixture was stirred for 1 h at 25°C, filtered, extracted with ethyl acetate. After removal of the solvent, N-Bocoligopeptide methyl ester 2 was isolated as a diastereomeric mixture by column chromatography on silica gel (hexane/ethyl acetate).

Acknowledgements

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- The chemical shifts (δ) of methyl protons (doublet, J=7.0– 7.3 Hz) at the N-terminal alanine residues in 2a–d and 4a–d were as follows: (S)-2a 1.31; (R)-2a 1.29; (S)-2b 1.25; (R)-2b

1.21; (S)-2c 1.21; (R)-2c 1.18; (S)-2d 1.20; (R)-2d 1.16; (S)-4a 1.355; (R)-4a 1.364; (S)-4b 1.33; (R)-4b 1.35; (S)-4c 1.31; (R)-4c 1.34; (S)-4d 1.32; (R)-4d 1.33. The chemical shifts (δ) of Boc protons (singlet) in 2e-h and 4e-h were as follows: (S)-2e 1.40; (R)-2e 1.38; (S)-2f 1.37; (R)-2f 1.39; (S)-2g 1.34; (R)-2g 1.39; (S)-2h 1.30; (R)-2h 1.34; (S)-4e 1.44; (R)-4e 1.45; (S)-4f 1.44; (R)-4f 1.44; (S)-4g 1.45; (R)-4g 1.43; (S)-4h 1.47; (R)-4h 1.43. The chemical shifts (δ) of methoxy protons (singlet) in 4f were as follows: (S)-4f 3.722; (R)-4f 3.716.

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